

A Convenient Approach for Access to Both Carbapentofuranoses and Carbahexopyranoses. Stereocontrolled Synthesis of Enantiopure Carba-D-ribofuranoses, Carba-D-arabinofuranoses and Carba-L-gulopyranose

Subrata Ghosh,* Tanurima Bhaumik, Niladri Sarkar, and Abhijit Nayek

Department of Organic Chemistry, Indian Association for the Cultivation of Science, Jadavpur, Kolkata 700 032, India

ocsg@iacs.res.in

Received August 18, 2006



A new approach to carbasugars in enantiomerically pure form is reported. The key step involves ringclosing metathesis of dienols 6 derived from a (*R*)-(+)-glyceraldehyde derivative 4 to form the substituted cyclopentenol 9 and cyclohexenol 34a. Stereocontrolled addition of hydroxyl groups followed by conversion of the ketal unit to hydroxymethyl group in these intermediates led to carbapentoses and -hexoses. Stereoselectivity during introduction of hydroxyl groups arises through the steric hindrance posed by the allylic substituents. A remarkable feature of the present approach is the accessibility of both D- and L-series of carbapentoses as illustrated by the synthesis of β -D- and β -L-carbaribofuranoses 17 and 20, respectively. Carba- α -D-ribofuranose 25, the biosynthetic intermediate to the antibiotic aristeromycin, has also been synthesized from the same cyclopentenol 9. Functional group manipulation in the cyclopentenol 9a also enabled access to carbaarabinofuranose 32. The present synthetic strategy can be extended for the synthesis of carbahexopyranose, as illustrated by the synthesis of carba- β -Lgulopyranose 40b.

Introduction

Carbocyclic analogues of pentoses and hexoses, often referred to as carbasugars,¹ are structural features present in many biologically interesting molecules. Because of improved acid and metabolic stability relative to their glycoside counterparts, compounds containing carbasugar moieties are potential therapeutic agents.² For example, a number of carbasugar-containing nucleoside analogues exhibits promising antiviral activities. Also oligosaccharide analogues containing carbasugar residues have been reported to be efficient glycosyltransferase inhibitors³ which have considerable therapeutic potential in the management of cancer, diabetes, and viral infections. Thus, development of efficient routes for the synthesis of carbasugars continues to be at the forefront of organic synthesis.^{4,5}

Apart from the construction of highly oxygenated carbocycles, stereocontrol in the synthesis of carbasugars is the major hurdle. Synthetic investigations made so far toward carbapentoses involve either transformation of carbohydrates^{4a-j} with hydroxyl groups of predefined stereochemistry or fragmentation of norbornene derivatives.^{4k-n} A few approaches involving construction of cyclopentene derivatives⁵ followed by hydroxylation have also been reported. A novel substrate-dependent dihydroxylation of cyclopentene has been reported recently by Miller

^{(1) (}a) Agrofoglio, L.; Suhas, E.; Farese, A.; Condom, R.; Challand, S. R.; Earl, R. A.; Guedj, R. *Tetrahedron* **1994**, *50*, 10611. (b) Suami, T. *Top. Curr. Chem.* **1990**, *154*, 257.

^{(2) (}a) Crimmins, M. T. *Tetrahedron* **1998**, *54*, 9229. (b) De Clercq, E. *Nucleosides Nucleotides* **1998**, *17*, 625. (c) Ferrero, M.; Gotor, V. *Chem. Rev.* **2000**, *100*, 4319.

^{(3) (}a) Ogawa, S.; Matsunaga, N.; Li, H.; Palcic, M. M. *Eur. J. Org. Chem.* **1999**, *3*, 631. (b) Ogawa, S.; Furuya, T.; Tsunoda, H.; Hindsgaul, O.; Stangier, K.; Palcic, M. M. *Carbohydr. Res.* **1995**, *271*, 197.



and co-workers.⁶ All of these investigations were mainly aimed either at carbapentoses or at carbapyranoses, but there was no general approach that can lead access to both carbapentoses and carbahexoses.⁷

As part of our continued interest⁸ in the synthesis of cyclopentane-containing molecules of biological interest, we initiated a program to design a general approach that would allow access to both carbapentofuranoses and carbahexopyranoses. We anticipated that carbaanalogues of pentoses and hexoses could be prepared from the carbocyclic enols of the general structure 1 (Scheme 1) on stereocontrolled introduction of the hydroxyl groups. The ketal moiety at the allylic center would be the source of chirality as well as the hydroxymethyl group. The cyclic alkenols 1 would be available by ring-closing metathesis 9,10 of the dienols **2**, which in turn would be available from the unsaturated aldehyde 3 on addition of alkenyl metal having the required number of methylenes for entry into either carbapentoses or carbahexoses. The unsaturated aldehyde 3 would be available from (R)-(+)-glyceraldehyde derivative 4. We herein report the results of our investigation based on this concept.

(5) (a) Shoberu, K. A.; Roberts, S. M. *J. Chem. Soc., Perkin Trans. 1* **1992**, 2419. (b) Parry, R. J.; Haridas, K.; De Jong, R.; Johnson, C. R. *Tetrahedron Lett.* **1990**, *31*, 7549.

Results and Discussion

For entry into carbapentoses, cyclopentenol 9a (Scheme 2) is required. Toward this end the (R)-(+)-glyceraldehyde derivative 4 was converted to a chromatographically separable 1:1 mixture of the unsaturated esters 5a and 5b following the literature procedure.^{10e} Conversion of the pure ester **5a** to the aldehyde 3a and reaction of the latter with vinyl magnesium bromide gave a 1:1 mixture of the dienols 6a. Ring-closing metathesis of the dienols 6a with Grubbs I Ru-catalyst (PCy₃)₂- Cl_2Ru =CHPh afforded a mixture of the cyclopentenols **7a**.^{10f} The desired cyclopentenol 9a could however be obtained predominantly in the following way. We anticipated that reduction of the cyclopentenone, to be available from oxidation of this mixture of cyclopentenols, would proceed from the side opposite to the C-4 substituent to provide the desired cyclopentenol predominantly. The mixture of the cyclopentenols 7a was oxidized to give the ketone 8a in 96% yield. Reduction of the ketone 8a with LiAlH₄ gave a mixture of the cyclopentenols 9a along with its C-1 diastereomer in 16:1 ratio. The major cyclopentenol 9a was obtained in 79% yield after column chromatography. The stereochemical assignment to 9a was based on the NOE (1.8%) between C1 and C4 hydrogens. In a similar fashion the aldehyde 3b, prepared from the diastereoisomeric ester 5b, was converted to the cyclopentenol 9b.

After successful synthesis of the cyclopentenols 9a and 9b, we focused on their transformation to both β -D- and β -Lcarbaribofuranoses. The hydroxyl group in 9a and 9b was protected to provide the benzyl ethers **10a** and **10b**, respectively. Dihydroxylation of the cyclopentene derivative 10a was achieved with OsO_4 to produce exclusively the diol **11** in 90% yield. Stereochemical assignment to the diol 11 was based on Kishi's principle,¹¹ according to which hydroxylation took place from the face of the double bond opposite to the alkoxy group. This stereochemical assignment was further confirmed through transformation of the diol 11 to carba- β -D-ribofuranose 17 (vide infra). For completion of the synthesis, the diol 11 was protected to afford the tribenzyl derivative 12. Acid-induced deketalization of 12 afforded the diol 13. Periodate cleavage of the diol 13 followed by reduction of the resulting aldehyde 15 afforded the polyoxygenated cyclopentane 16 in overall excellent yield. Finally, hydrogenolysis led to carba- β -D-ribofuranose 17, $[\alpha]_D$ +8.5 (c 1.7, MeOH) [lit.^{4k} $[\alpha]_D$ +8.0 (c 2.9, MeOH)]. ¹H and ¹³C NMR spectra of the sample obtained in this way were closely comparable to those reported in literature.⁴ⁱ Following a similar protocol, dihydroxylation of the cyclopentene 10b

^{(4) (}a) Tadano, K.; Maeda, H.; Hoshino, M.; Iimura, Y.; Suami, T. J. Org. Chem. 1987, 52, 1946. (b) Rassu, G.; Auzzas, L.; Pinna, L.; Zambrano, V.; Battistini, L.; Zanurdi, F.; Marzocchi, L.; Acquotti, D.; Casiraghi, G. J. Org. Chem. 2001, 66, 8070. (c) Tadano, K.; Maeda, H.; Hoshino, M.; Iimura, Y.; Suami, T. Chem. Lett. 1986, 1081. (d) Callam, C. S.; Lowary, T. L. Org. Lett. 2000, 2, 167. (e) Horneman, A. M.; Lundt, I. J. Org. Chem. 1998, 63, 1919. (f) Callam, C. S.; Lowary, T. L. J. Org. Chem. 2001, 66, 8961. (g) Tadano, K.; Kimura, H.; Hoshino, M.; Ogawa, S.; Suami, T. Bull. Chem. Soc. Jpn. 1987, 60, 3673. (h) Tadano, K.; Hoshino, M.; Ogawa, S.; Suami, T. J. Org. Chem. **1988**, 53, 1427. (i) Tadano, K.; Hakuba, K.; Kimura, H.; Ogawa, S. J. Org. Chem. **1989**, 54, 276. (j) Yoshikawa, M.; Yokokawa, Y.; Inoue, Y.; Yamaguchi, S.; Murukami, N.; Kitagawa, I. Tetrahedron 1994, 50, 9961. (k) Marschner, C.; Penn, G.; Griengl, H. Tetrahedron Lett. 1990, 31, 2873. (1) Mehta, G.; Reddy, D. S. Tetrahedron Lett. 1999, 40, 9137. (m) Mehta, G.; Mohal, N. Tetrahedron Lett. 1999, 40, 5791. (n) Marschner, C.; Baumgartner, J.; Griengl, H. J. Org. Chem. 1995, 60, 5224.

⁽⁶⁾ Jiang, M. X. -W.; Jin, B.; Gage, J. L.; Priour, A.; Savela, G.; Miller, M. J. *J. Org. Chem.* **2006**, *71*, 4164 and references therein.

^{(7) (}a) McCasland, G. E.; Furuta, S.; Durham, L. J. J. Org. Chem. 1966, 31, 1516. (b) Takahashi, T.; Kotsubo, H.; Iyobe, A.; Namiki, T.; Koizumi, T. J. Chem. Soc., Perkin Trans. 1 1990, 3065. (c) Shing, T. K. M.; Cui, Y.; Tang, Y. J. Chem. Soc., Chem. Commun. 1991, 756. (d) Dumortier, L.; Vander Eycken, J.; Vandewalle, M. Synlett 1992, 245. (e) Mehta, G.; Mohal, N. Tetrahedron Lett. 1998, 39, 3285. (f) Pingli, L.; Vandewalle, M. Synlett 1994, 228.

^{(8) (}a) Ghosh, S.; Patra, D.; Saha, G. J. Chem. Soc., Chem. Commun.
1993, 783. (b) Ghosh, S.; Patra, D. Tetrahedron Lett. 1993, 34, 4565. (c) Patra, D.; Ghosh, S. J. Org. Chem. 1995, 60, 2526. (d) Haque, A.; Ghatak, A.; Ghosh, S.; Ghoshal, N. J. Org. Chem. 1997, 62, 5211. (e) Samajdar, S.; Patra, D.; Ghosh, S. Tetrahedron 1998, 54, 1789. (f) Samajdar, S.; Ghatak, A.; Ghosh, S. Tetrahedron Lett. 1999, 40, 4401. (g) Banerjee, S.; Ghosh, S. J. Org. Chem. 2003, 68, 3981. (h) Sarkar, N.; Nayek, A.; Ghosh, S. Org. Lett. 2004, 6, 1903.

⁽⁹⁾ For reviews on RCM reaction, see: (a) Schuster, M.; Blechert, S. Angew. Chem., Int. Ed. Engl. 1997, 36, 2036. (b) Grubbs, R. H.; Chang, S. Tetrahedron 1998, 54, 4413. (c) Pariya, C.; Jayaprakash, K. N.; Sarkar, A. Coord. Chem. Rev. 1998, 168, 1. (d) Frustner, A. Angew. Chem., Int. Ed. 2000, 39, 3012. (e) Trnka, T. M.; Grubbs, R. H. Acc. Chem. Res. 2001, 34, 18. (f) Kotha, S.; Sreenivasachary, N. Indian J. Chem. 2001, 40B, 763. (g) Grubbs, R. H. Tetrahedron 2004, 60, 7117. (h) Dieters, A.; Martin, S. F. Chem. Rev. 2004, 104, 2199. (l) Nicolaou, K. C.; Bulger, P. G.; Sarlah, D. Angew. Chem., Int. Ed. 2005, 44, 4490. (j) Ghosh, S.; Ghosh, S.; Sarkar, N. J. Chem. Sci 2006, 118, 223.

⁽¹⁰⁾ For our work on RCM reaction, see: (a) Holt, D. J.; Barker, W. D.; Jenkins, P. R.; Davies, D. L.; Garratt, S.; Fawcett, J.; Rusell, D. R.; Ghosh, S. Angew. Chem., Int. Ed. 1998, 37, 3298. (b) Haque, A.; Panda, J.; Ghosh, S. Indian J. Chem. 1999, 38B, 8. (c) Holt, D. J.; Barker, W. D.; Jenkins, P. R.; Panda, J.; Ghosh, S. J. Org. Chem. 2000, 65, 482. (d) Nayek, A.; Banerjee, S.; Sinha, S.; Ghosh, S. Tetrahedron Lett. 2004, 45, 6457. (e) Banerjee, S.; Ghosh, S.; Sinha, S.; Ghosh, S. J. Org. Chem. 2005, 70, 4199. (f) Nayek, A.; Sinha, S.; Bhaumik, T.; Ghosh, S. J. Mol. Catal. A: Chem. 2006, 254, 85. (g) Ghosh, S.; Sinha, S.; Drew, M. G. B. Org. Lett. 2006, 8, 3781.

⁽¹¹⁾ Cha, J. K.; Christ, W. J.; Kishi, Y. Tetrahedron 1984, 40, 2247.

JOC Article

SCHEME 2^a



^{*a*} For structures **3**–12, 18, and 19: R^1 , $R^2 = -(CH_2)_{5^-}$ and for structures **3** and **5**–8: **a**, $R^3 = \alpha$ -H; **b**, $R^3 = \beta$ -H.

afforded the diol **18**, which on benzylation gave the tribenzyl derivative **19**. Carba- β -L-ribofuranose **20**, $[\alpha]_D$ -6.2 (*c* 1.5, MeOH), was then obtained from **19** following the protocol used for the synthesis of **17**. Thus, availability of the carbaribofuranoses of both the D- and L-series from a single enantiomer, (*R*)-(+)-glyceraldehyde derivative **4** is a remarkable advantage of this approach. Further, hydrogenolysis of the tribenzyl derivative **13** afforded the carbaaldohexofuranose **14**, a new class of carbasugars for synthesis of which there is only a single report.⁴

The cyclopentenol **9a** can also be employed for synthesis of the carba- α -D-ribofuranose. For this purpose it is necessary to invert the configuration of the C-1 hydroxyl group. This was achieved as delineated in Scheme 3. The cyclopentenol **9a** was converted to the silyl ether **21**. Dihydroxylation of the cyclopentene **21** followed by benzylation of the diol **22a** gave the dibenzyl ether **22b**. The latter on desilylation gave the cyclopentanol **22c**. Oxidation of the cyclopentanol **22c** with Jones reagent afforded the ketone **23** in 87% yield. Reduction of the ketone **23** with LiAlH₄ took place from addition of the hydride predominantly from the side opposite to C-2, C-3 benzyloxy groups to afford the cyclopentanol **24** in 60% isolated yield with 9% yield of the diastereomeric cyclopentanol **22c**. The ketal group in **24** was then converted to a hydroxymethyl group using the four-step protocol for conversion of **12** to **17** as described in Scheme 2 to afford carba- α -D-ribofuranose **25**, $[\alpha]_D$ +42.4 (*c* 1.7, MeOH) [lit.^{5b} $[\alpha]_D$ +46.1 (*c* 0.8, MeOH)]. ¹H NMR^{4h} and ¹³C NMR^{5b} spectral data of the carbasugar **25** were closely comparable to those reported in literature. The importance of the carbasugar **25** as a biosynthetic intermediate to the antibiotic aristeromycin **26**^{5b} has already been established.

The cyclopentanol **22c** can be used for entry into the arabino series. It is interesting to note that whereas Jones oxidation of the cyclopentanol **22c** gave the cyclopentanone **23**, Swern oxidation gave the cyclopentenone **27** in excellent yield (Scheme 4). Triethylamine used during Swern oxidation probably facilitated elimination of the benzyloxy group β to the carbonyl of the initially formed cyclopentanone **23**. Hydroboration of the enone **27** gave a mixture of the diols, which on chromatography afforded the pure diols **29a** (20%) and **30a** (54%). The moderate stereoselectivity observed during this reaction may be rationalized as follows. Reduction of carbonyl group in conjugated enone with diborane generally proceeds¹² at a faster rate to form

⁽¹²⁾ Stefanovic, M.; Lajsic, S. Tetrahedron Lett. 1967, 8, 1777.

SCHEME 3^a



^{*a*} For structures **21–24**: R^1 , $R^2 = -(CH_2)_5$ -.

SCHEME 4^a



^{*a*}For structures **27–30**: R^1 , $R^2 = -(CH_2)_5$ -.

allylic alcohol. Thus reaction of the enone **27** probably proceeded through preferential addition of diborane to an initially formed carbonyl-borane complex from the face opposite to the C-4 substituent leading to a mixture of the nonisolable

allylic alcohols **28a** and **28b**, with the latter predominating. Subsequent addition of borane to the alkene units in these alcohols occurred exclusively from the opposite face of the C-4 substituent leading to the diols **29a** and **30a**. For completion of the synthesis of the carbaarabinofuranoses, the diols **29a** and **30a** were benzylated to afford **29b** and **30b**, respectively. Each of the benzyloxy cyclopentanes **29b** and **30b** were subjected to the four-step protocol already described for conversion of **12** to carba- β -D-ribofuranose **17**, to give carba- α -D-arabinofuranose **31**, $[\alpha]_D + 44.5$ (*c* 1.5, MeOH) [lit.⁴ⁱ (for *ent*-**31**) $[\alpha]_D - 40.5$ (*c* 0.84, MeOH)] and carba- β -D-arabinofuranose **32**, $[\alpha]_D + 9.2$ (*c* 1.5, MeOH) [lit.^{4g} $[\alpha]_D + 9.7$ (*c* 0.7, MeOH)]. ¹H and ¹³C NMR spectral data of the carbasugars **31** and **32** were also comparable to those reported in literature.^{4i,n}

The above protocol can be extended for access to carbahexopyranoses as illustrated by the synthesis of 5a-carba- β -Lgulopyranose. Toward this end, the aldehyde 3b was allowed to react with allyl zinc to afford a diastereoisomeric mixture of the alcohols 33 (Scheme 5). Ring-closing metathesis of this dienol mixture with Grubbs' first generation catalyst afforded the cyclohexenols 34a and 34b in excellent yield. The cyclohexenol 34a could be exclusively obtained by oxidation of the mixture of the cyclohexenols 34a and 34b to the cyclohexenone 34c followed by its reduction with LiAlH₄. NOE (1.8%) between C-1 and C-5 hydrogens in the acetate 34d derived from 34a confirmed the syn stereochemical assignment. For introduction of the hydroxyl groups, the cyclohexenol 34a was subjected to epoxidation with m-CPBA. A mixture of products was obtained from which no desired epoxide could be isolated. However, epoxidation of the acetate 34d with m-CPBA proceeded smoothly to produce in 89% yield an inseparable mixture of the oxiranes 35 and the diastereoisomer 36. An attempt to convert these oxiranes to the corresponding allylic alcohols using the procedure (Ph2Se2/NaBH4/30% H2O2) of Sharpless and Lauer¹³ led to isolation of the ene-diol **38a** in 41% yield. During this reaction, the oxirane ring in 36 remained unchanged and produced only the deacetylated epoxide 37 in 29% yield. The inertness of the oxirane ring in 36 toward nucleophilic opening may be explained as follows. Opening of the oxirane ring in 36 requires the nucleophile to approach from the opposite face of the epoxide ring, which is hindered by the bulky ketal substituent. The hydroxyl groups in the cyclohexenediol 38a were protected to provide the benzyl ether 38b. Dihydroxylation with OsO₄ led exclusively the cis-diol 39a. The stereochemical assignment to the diol 39a follows from its conversion to the known gulopyranose 40b as stated below. The diol 39a was benzylated, and the ketal unit in tetrabenzylated derivative 39b was converted to the hydroxymethyl group using the four-step protocol used in the synthesis of carbafuranoses to afford carba- β -L-gulopyranose **40b**. The pentaacetate **40c**, $[\alpha]_D^{25} + 21.5$ (c 2.0, CHCl₃) [lit.^{7f} $[\alpha]^{20}_{D}$ +20.5 (c 1.0, CHCl₃)] obtained from acetylation of 40b displayed ¹H NMR spectral data comparable to those reported in literature.7f

Conclusion

We have developed a new synthetic approach that provided access to both carbapentoses and carbahexoses in enantiomerically pure form. A ring-closing metathesis of dienols derived from (R)-(+)-glyceraldehyde derivative afforded the basic carbocyclic units, which on stereoselective hydroxylation gave

⁽¹³⁾ Sharpless, K. B.; Lauer, R. F. J. Am. Chem. Soc. 1973, 95, 2697.

JOC Article

SCHEME 5^a



^{*a*} For structures **3b** and **33–39**: R^1 , $R^2 = -(CH_2)_5$ -.

the carbasugars. The synthesis of both the D- and L-series of carbaribofuranoses from a single enantiomer of glyceraldehyde derivative is the most remarkable feature of the present approach.

Experimental Section

(4S)-4-[(2S)-1,4-Dioxaspiro[4.5]dec-2-yl]cyclopent-2-en-1one 8a. To a magnetically stirred solution of the mixture of the cyclopentenol derivatives 7a, prepared by ring-closing metathesis of the dienol mixture 6a according to the previously published procedure, 10f (350 mg, 1.56 mmol) in acetone (10 mL) was added dropwise Jones reagent (1.2 mL) at 0 °C till the color of the reagent persisted. After stirring at that temperature for an additional 30 min, the reaction mixture was diluted with water (2 mL) and extracted with diethyl ether (2 \times 15 mL). The combined organic layer was washed sequentially with aqueous NaOH solution (20%, 5×1 mL) and water $(5 \times 1 \text{ mL})$ and dried. Removal of the solvent in vacuo followed by column chromatography using ether/petroleum ether (1:6) as eluent afforded the cyclopentenone derivative 8a (333 mg, 96%): IR 1715.0 cm⁻¹; ¹H NMR δ 1.41 (2H, br s, CH₂), 1.53– 1.64 (8H, m, CH₂), 2.04 (1H, dd, J = 2.4, 18.6 Hz, CH₂), 2.45 $(1H, dd, J = 6.6, 18.9 Hz, CH_2), 3.08-3.14 (1H, m, CH), 3.67-$ 3.73 (1H, m), 3.98-4.09 (2H, m), 6.26 (1H, dd, J = 1.8, 5.4 Hz)=CH), 7.77 (1H, dd, J = 2.4, 5.4 Hz, =CH); ¹³C NMR δ 23.7 (CH₂), 23.9 (CH₂), 25.0 (CH₂), 34.6 (CH₂), 36.4 (CH₂), 36.9 (CH₂), 45.4 (CH), 67.2 (OCH₂), 77.3 (OCH), 110.3 (C), 135.3 (CH), 165.1 (CH), 208.5 (CO); HRMS (ESI) calcd for $C_{13}H_{18}O_3Na (M + Na)^+$, 245.1154; found, 245.1112.

Reduction of Ketone 8a. (1R,4S)-4-[(2S)-1,4-Dioxaspiro[4.5]dec-2-yl]cyclopent-2-en-1-ol 9a and its C-1 Diastereoisomer. To a stirred suspension of LiAlH₄ (260 mg, 6.84 mmol) in diethyl ether (15 mL) at -60 °C was added dropwise a solution of the cyclopentenone derivative 8a (1.38 g, 6.22 mmol) in diethyl ether (20 mL). After stirring at that temperature for 1 h, the reaction mixture was quenched by sequential addition of water (0.26 mL), aqueous NaOH solution (15%, 0.26 mL), and water (0.78 mL) and allowed to attain room temperature. After stirring at room temperature for 15 min, the white solid formed was filtered off. The solid was washed thoroughly with diethyl ether. The combined ether layer was dried. Removal of solvent in vacuo followed by column chromatography of the residual mass with silica gel (100–200 mesh) using ether/petroleum ether (1:3) as eluent afforded the cyclopentenol derivatives **9a** (1.1 g, 79%) and its C-1 diastereomer (70 mg, 5%) as viscous liquid.

9a. $R_f = 0.54$ (EtOAc/petroleum ether 1:1); $[\alpha]^{22}{}_D -90.5$ (*c* 5.1, CHCl₃); ¹H NMR δ 1.37 (2H, br s, CH₂), 1.56–1.57 (8H, m, CH₂), 2.26–2.36 (2H, m, CH₂), 2.73–2.77 (2H, m), 3.70 (1H, dd, *J* = 6.5, 8.0 Hz, OCH₂), 4.04 (1H, dd, *J* = 6.3, 7.7 Hz, OCH₂), 4.10–4.15 (1H, m, OCH), 4.62–4.64 (1H, m, C₁–H), 5.88 (1H, dd, *J* = 2.2, 5.5 Hz, =CH), 6.01–6.02 (1H, m, =CH); ¹³C NMR δ 24.1 (CH₂), 24.3 (CH₂), 25.4 (CH₂), 35.1 (CH₂), 36.2 (CH₂), 37.5 (CH₂), 47.0 (CH), 67.4 (OCH₂), 75.8 (OCH), 77.8 (OCH), 110.2 (C), 133.1 (CH), 136.8 (CH); HRMS (ESI) calcd for C₁₃H₂₀O₃Na (M + Na)⁺, 247.1310; found, 247.1343.

C-1 Diastereomer of 9a. $R_f = 0.49$ (EtOAc/petroleum ether 1:1); [α]²²_D -161.7 (*c* 2.9, CHCl₃); ¹H NMR δ 1.40 (2H, br s, CH₂), 1.52-1.68 (8H, m, CH₂), 1.75-1.93 (2H, m, CH₂), 3.03-3.09 (2H, m), 3.58 (1H, t, J = 6.9 Hz), 3.87 (1H, dd, J = 6.4, 13.0 Hz), 3.97-4.02 (1H, m), 4.87-4.90 (1H, m, C₁-H), 5.92-5.94 (1H, m, =CH), 6.05-6.08 (1H, m, =CH); ¹³C NMR δ 23.8 (CH₂), 24.0 (CH₂), 25.1 (CH₂), 34.9 (CH₂), 36.1 (CH₂), 36.4 (CH₂), 48.3 (CH), 67.5 (OCH₂), 76.9 (OCH), 78.9 (OCH), 109.7 (C), 134.6 (CH), 136.5 (CH); HRMS (ESI) calcd for C₁₃H₂₀O₃Na (M + Na)⁺, 247.1310; found, 247.1313.

(2S)-2-[(1S,4R)-4-(Benzyloxy)cyclopent-2-en-1-yl]-1,4dioxaspiro[4.5]decane 10a. To a magnetically stirred suspension of NaH (60 mg, 1.25 mmol, 50% in oil), freed from adhering oil

by repeated washing with petroleum, in dry THF (3 mL) at 0 °C was added dropwise a solution of the alcohol 9a (140 mg, 0.63 mmol) in dry THF (3 mL) under N2 atmosphere. The mixture was allowed to stir at room temperature for 2 h, and then to it was added HMPA (0.3 mL) dropwise followed by benzyl bromide (0.15 mL, 1.25 mmol). After stirring for an additional 12 h at room temperature, the reaction mixture was cooled to 0 °C and quenched by adding cold water (1 mL). Usual workup of the reaction mixture followed by column chromatography using ether/petroleum ether (1:4) as eluent afforded the benzyl ether **10a** (180 mg, 92%) as a colorless viscous liquid: $[\alpha]^{23}_{D}$ –23.9 (c 1.5, CHCl₃); ¹H NMR δ 1.39-1.49 (2H, m, CH₂), 1.56-1.63 (8H, m, CH₂), 2.27 (2H, td, J = 7.7, 13.7 Hz, CH₂), 2.74–2.77 (1H, m, CH), 3.66 (1H, ddd, J = 3.4, 5.9, 9.3 Hz), 3.94-4.04 (3H, m), 4.51 (1H, d, J = 11.9 Hz, PhCH₂), 4.56 (1H, d, J = 12.0 Hz, PhCH₂), 5.94 (1H, td, J = 2.0, 5.7 Hz, =CH), 6.03 (1H, td, J = 1.7, 5.8 Hz, =CH), 7.25-7.34 (5H, m, ArH); ^{13}C NMR δ 23.8 (CH₂), 24.0 (CH₂), 25.1 (CH₂), 32.2 (CH₂), 34.9 (CH₂), 36.5 (CH₂), 48.0 (CH), 67.3 (OCH₂), 70.9 (OCH₂) 78.8 (OCH), 83.7 (OCH), 109.5 (C), 127.5 (CH), 127.7 (CH), 128.3 (CH), 132.4 (CH), 135.9 (CH), 138.5 (C); HRMS (ESI) calcd for $C_{20}H_{26}O_3Na$ (M + Na)⁺, 337.1780; found, 337.1784.

(1R,2R,3R,5S)-3-(Benzyloxy)-5-[(2S)-1,4-dioxaspiro[4.5]dec-2-yl]cyclopentane-1,2-diol 11. To a magnetically stirred solution of the cyclopentene derivative 10a (160 mg, 0.51 mmol) in acetone/ water (4:1, 7.5 mL) at room temperature was added N-methylmorpholine N-oxide monohydrate (207 mg, 1.53 mmol) followed by a catalytic amount of OsO4. After the reaction mixture stirred at room temperature for 14 h, acetone was removed under reduced pressure. The residual mass was extracted with chloroform $(2 \times 10 \text{ mL})$, washed with brine $(2 \times 2 \text{ mL})$, and dried. Removal of solvent in vacuo followed by column chromatography using ethyl acetate/ petroleum ether (2:3) afforded the *cis*-diol **11** (159 mg, 90%): $[\alpha]^{23}_{D}$ +1.2 (*c* 1.5, CHCl₃); ¹H NMR δ 1.26 (1H, ddd, *J* = 5.6, 9.6, 13.1 Hz, CH₂), 1.39 (2H, br s, CH₂), 1.57-1.61 (8H, m, CH₂), 1.98-2.09 (1H, m, CH), 2.17 (1H, td, J = 8.3, 13.3 Hz, CH₂), 3.13 (1H, br s, OH), 3.32 (1H, br s, OH), 3.60 (1H, dd, J = 6.0, 7.0 Hz), 3.86 (1H, dt, J = 2.0, 7.4 Hz), 4.0-4.12 (4H, m), 4.51 (1H, d, J = 11.8 Hz, PhCH₂), 4.57 (1H, d, J = 11.8 Hz, PhCH₂), 7.24-7.38 (5H, m, ArH); ¹³C NMR δ 23.7 (CH₂), 23.9 (CH₂), 24.9 (CH₂), 30.4 (CH₂), 34.8 (CH₂), 36.2 (CH₂), 45.1 (CH), 67.9 (OCH₂), 71.5 (OCH₂), 75.4 (OCH), 75.8 (OCH), 78.9 (OCH), 82.7 (OCH), 109.7 (C), 127.5 (CH), 128.2 (CH), 138.0 (C); HRMS (ESI) calcd for $C_{20}H_{28}O_5Na (M + Na)^+$, 371.1834; found, 371.1840.

(2S)-2-[(1R,2R,3S,4R)-2,3,4-tris(Benzyloxy)cyclopent-1-yl]-1,4dioxaspiro[4.5]decane 12. Following the procedure for the benzylation of the alcohol 9a, the *cis*-diol 11 (140 mg, 0.40 mmol) was benzylated with benzyl bromide (0.19 mL, 1.61 mmol) to afford the benzyl ether **12** (180 mg, 85%) as a colorless viscous liquid: $[\alpha]^{23}_{D} - 16.8 (c \ 1.5, CHCl_3); {}^{1}H \ NMR \ \delta \ 1.22 - 1.30 (1H, m, CH_2),$ 1.39 (2H, br s, CH₂), 1.49–1.61 (8H, m, CH₂), 2.17–2.33 (2H, m, CH₂, CH), 3.62 (1H, dd, J = 6.6, 6.9 Hz), 3.81 (1H, t, J = 5.4Hz), 3.92-4.02 (3H, m), 4.13 (1H, dd, J = 6.6, 13.2 Hz), 4.52-4.66 (6H, m, PhCH₂), 7.23–7.39 (15H, m, ArH); ¹³C NMR δ 23.7 (CH₂), 23.9 (CH₂), 25.1 (CH₂), 29.5 (CH₂), 34.8 (CH₂), 36.4 (CH₂), 43.7 (CH), 67.2 (OCH₂), 71.2 (OCH₂), 71.6 (OCH₂), 71.7 (OCH₂), 77.1 (OCH), 78.3 (OCH), 81.0 (OCH), 82.9 (OCH), 109.5 (C), 127.36 (CH), 127.40 (CH), 127.43 (CH), 127.5 (CH), 127.6 (CH), 128.0 (CH), 128.12 (CH), 128.14 (CH), 128.2 (CH), 138.3 (C), 138.4 (C); HRMS (ESI) calcd for $C_{34}H_{41}O_5$ (M + H)⁺, 529.2954; found, 529.2917.

(15)-1-[(1R,2R,3S,4R)-2,3-4-Tris(benzyloxy)cyclopent-1-yl]ethane-1,2-diol 13. A mixture of the cyclopentane derivative 12 (160 mg, 0.30 mmol) and aqueous acetic acid (80%, 2.5 mL) was stirred at room temperature for 17 h. The resulting solution was diluted with ethyl acetate (10 mL) and washed repeatedly with aqueous NaOH solution (20%, 5 × 1.5 mL) to make it just alkaline (pH paper). The organic layer was washed with brine (2 × 1 mL), dried, and concentrated, and the residual mass was column chromatographed using ethyl acetate/petroleum ether (2:3) to afford the diol **13** (96 mg, 71%): $[\alpha]^{24}_{D}$ +44.7 (*c* 1.5, CHCl₃); ¹H NMR δ 1.33 (1H, ddd, *J* = 4.2, 8.5, 13.0 Hz, CH₂), 2.21–2.31 (1H, m, CH), 2.37 (1H, t, *J* = 8.3 Hz, CH₂), 2.75 (2H, br s, OH), 3.45 (1H, dd, *J* = 5.7, 11.7 Hz, OCH₂), 3.59 (2H, dd, *J* = 3.0, 11.7 Hz, OCH₂, OCH), 3.88 (1H, dd, *J* = 1.9, 4.5 Hz), 3.95 (2H, dd, *J* = 4.7, 7.7 Hz), 4.39–4.60 (6H, m, PhCH₂), 7.29–7.34 (15H, m, ArH); ¹³C NMR δ 30.5 (CH₂), 42.1 (CH), 65.1 (OCH₂), 71.25 (OCH₂), 71.27 (OCH₂), 71.7 (OCH₂), 75.0 (OCH), 80.0 (OCH), 80.1 (OCH), 81.3 (OCH), 127.5 (CH), 127.70 (CH), 127.72 (CH), 127.8 (CH), 127.9 (CH), 128.0 (CH), 128.3 (CH), 128.37 (CH), 128.38 (CH), 137.3 (C), 137.9 (C), 137.9 (C); HRMS (ESI) calcd for C₂₈H₃₃O₅ (M + H)⁺, 449.2328; found, 449.2328.

(1*R*,2*S*,3*R*,4*R*)-4-[(1*S*)-1,2-Dihydroxyethyl]cyclopentane-1,2,3triol (Carba-β-D-allofuranose) 14. A solution of the diol 13 (16 mg, 0.03 mmol) in dry methanol (2 mL) containing Pd-C (10%) was stirred under hydrogen atmosphere at room temperature for 20 h. The catalyst was filtered off, and removal of solvent in vacuo afforded the alcohol 14 (6 mg, 94%): $[\alpha]^{24}_{D}$ -7.5 (*c* 0.5, CH₃-OH); ¹H NMR (CD₃OD) δ 1.25–1.33 (1H, m), 2.03 (1H, m), 2.12–2.22 (1H, m), 3.54–3.70 (4H, m), 3.96–3.98 (1H, m), 4.10 (1H, m); ¹³C NMR (CD₃OD) δ 33.7 (CH₂), 45.9 (CH), 66.0 (OCH₂), 73.7 (OCH), 75.5 (OCH), 76.5 (OCH), 79.6 (OCH); HRMS (ESI) calcd for C₇H₁₄O₅Na (M + Na)⁺, 201.0739; found, 201.0723.

(1S,2R,3S,4R)-2,3,4-Tris(benzyloxy)cyclopentanecarbaldehyde 15. To a magnetically stirred ice-cold solution of the diol 13 (80 mg, 0.18 mmol) in methanol/water (3:1, 1.6 mL) was added NaIO₄ (76 mg, 0.36 mmol) in multiple portions. The reaction mixture was allowed to stir at 0 °C for 30 min. The precipitated white solid was filtered off after washing it thoroughly with diethyl ether. Usual workup of the filtrate afforded the aldehyde 15 (70 mg, 94%): IR 1722.3 cm⁻¹; ¹H NMR δ 1.91–1.99 (1H, m, CH₂), 2.34 (1H, ddd, J = 6.6, 10.6, 14.3 Hz, CH₂), 3.05-3.12 (1H, m, CH), 3.89 (1H, t, J = 3.3 Hz), 4.03–4.06 (1H, m), 4.27 (1H, dd, J = 4.6, 6.4 Hz, $4.45 - 4.87 (6\text{H}, \text{m}, \text{PhCH}_2)$, 7.28 - 7.35 (15H, m, m)ArH), 9.72 (1H, s, CHO); ¹³C NMR δ 27.5 (CH₂), 53.3 (CH), 71.2 (OCH₂), 72.0 (OCH₂), 72.1 (OCH₂), 79.0 (OCH), 80.3 (OCH), 81.1 (OCH), 127.5 (CH), 127.61 (CH), 127.65 (CH), 127.70 (CH), 127.73 (CH), 128.28(CH), 128.31 (CH), 137.7 (C), 137.9 (C), 138.0 (C), 201.7 (CHO); HRMS (ESI) calcd for $C_{27}H_{28}O_4Na (M + Na)^+$, 439.1885; found 439.1832

[(1R,2R,3S,4R)-2,3,4-Tris(benzyloxy)cyclopent-1-yl]methanol 16. To a magnetically stirred ice-cold solution of the aldehyde 15 (70 mg, 0.17 mmol) in methanol (1.8 mL) under argon atmosphere was added NaBH₄ (13 mg, 0.34 mmol) in small portions, and the mixture was allowed to stir for 30 min at 0 °C. The reaction mixture was quenched with water. Usual workup of the residual mass obtained after evaporation of methanol under reduced pressure followed by column chromatography using ethyl acetate/petroleum ether (1:3) as eluent afforded the alcohol 16 (67 mg, 95%): $[\alpha]^{20}_{D}$ +33.0 (c1.5, CHCl₃); ¹H NMR δ 1.30 (1H, ddd, J = 4.9, 7.9, 13.1 Hz, CH₂), 2.21–2.31 (1H, m), 2.35–2.43 (2H, m), 3.56-3.63 (2H, m, OCH₂), 3.80 (1H, dd, J = 4.9, 7.1 Hz), 3.86 (1H, dd, J = 3.1, 4.7 Hz), 4.00 (1H, ddd, J = 3.1, 4.9, 7.6 Hz), 4.43–4.61 (6H, m, PhCH₂), 7.23–7.35 (15H, m, ArH); ¹³C NMR δ 30.1 (CH₂), 42.5 (CH), 65.1 (OCH₂), 71.3 (OCH₂), 71.5 (OCH₂), 71.7 (OCH₂), 80.8 (OCH), 81.07 (OCH), 81.09 (OCH), 127.5 (CH), 127.56 (CH), 127.58 (CH), 127.59 (CH), 127.78 (CH), 127.80 (CH), 128.2 (CH), 128.27 (CH), 128.31 (CH), 138.06 (C), 138.09 (C), 138.2 (C); HRMS (ESI) calcd for $C_{27}H_{30}O_4Na$ (M + Na)⁺, 441.2042; found, 441.2041.

Synthesis of Dienols 6b. To a magnetically stirred solution of vinyl magnesium bromide [prepared by addition of vinyl bromide (3.3 g, 31 mmol) in THF (40 mL) to magnesium turnings (370 mg, 15.5 mmol)] was added a solution of the aldehyde **3b** (690 mg, 3.1 mmol) in THF (8 mL) and refluxed gently for 1 h. The reaction mixture was allowed to attain room temperature, then cooled to 0 °C, quenched by water (4 mL), and filtered. Organic phase was washed with brine (3×4 mL) and dried. Removal of solvent in vacuo followed by column chromatography using ether/

petroleum ether (1:4) as eluent afforded the alcohol **6b** (550 mg, 70%) as a mixture of two diastereoisomers: IR 3445.0 cm⁻¹; ¹H NMR (of the diastereoisomeric mixture) δ 1.33 (4H, br s), 1.42–1.55 (18H, m), 1.74–1.87 (2H, m), 2.18–2.30 (1H, m), 2.33–2.46 (1H, m), 2.82 (2H, br s), 3.52–3.60 (2H, m), 3.84–3.91 (4H, m), 4.11–4.21 (2H, m), 4.99–5.20 (8H, m), 5.47–5.56 (2H, m), 5.77–5.85 (2H, m); ¹³C NMR (for one isomer from the mixture) δ 24.2 (CH₂), 24.2 (CH₂), 25.4 (CH₂), 35.5 (CH₂), 36.7 (CH₂), 39.8 (CH₂), 45.4 (CH), 68.2 (OCH₂), 71.4 (OCH), 78.3 (OCH), 110.3 (C), 115.4 (CH₂), 117.5 (CH₂), 138.3 (CH), 140.9 (CH) and (for the other isomer) δ 24.2 (CH₂), 24.2 (CH₂), 24.2 (CH₂), 25.4 (CH₂), 25.4 (CH₂), 35.5 (CH₂), 35.5 (CH₂), 36.7 (CH₂), 117.5 (CH₂), 117.7 (CH₂), 70.9 (OCH), 78.4 (OCH), 110.3 (C), 114.1 (CH₂), 117.7 (CH₂), 138.2 (CH), 141.9 (CH); HRMS (ESI) calcd for C₁₅H₂₄O₃Na (M + Na)⁺, 275.1623; found, 275.1627.

(1*S*,4*R*)-4-[(2*S*)-1,4-Dioxaspiro[4.5]dec-2-yl]cyclopent-2-en-1ol 9b and Its C-1 Diastereoisomer. A solution of the above diene mixture 6b (370 mg, 1.47 mmol) in dry dichloromethane(12 mL) was degassed by bubbling argon through it. To it was added Grubbs' first generation catalyst (44 mg, 0.05 mmol) in one portion. The resulting pink solution was stirred at room temperature under argon atmosphere for 24 h. The residual mass obtained after removal of solvent was chromatographed using ether/petroleum ether (1:3) as eluent to afford the pure cyclopentenol derivative 7b (280 mg, 85%) as a 1:1 diastereoisomeric mixture.

The mixture of the cyclopentenol derivatives **7b** (362 mg, 1.62 mmol) was oxidized with Jones reagent, as described for oxidation of **7a**, to the cyclopentenone derivative **8b** (310 mg, 86%): IR 1716.5 cm⁻¹; ¹H NMR δ 1.26–1.33 (2H, m, CH₂), 1.47–1.55 (8H, m, CH₂), 2.19 (1H, dd, J = 2.3, 18.8 Hz, CH₂), 2.39 (1H, dd, J = 6.5, 18.8 Hz, CH₂), 3.10–3.15 (1H, m, CH), 3.55 (1H, dd, J = 6.5, 8.2 Hz, OCH₂), 4.01 (1H, dd, J = 6.6, 8.2 Hz, OCH₂), 4.17 (1H, dd, J = 6.3, 11.6 Hz, OCH), 6.21 (1H, dd, J = 2.0, 5.6 Hz, =CH), 7.51 (1H, dd, J = 2.5, 5.7 Hz, =CH); ¹³C NMR δ 23.5 (CH₂), 23.7 (CH₂), 24.9 (CH₂), 34.3 (CH₂), 35.8 (CH₂), 36.3 (CH₂), 44.4 (CH), 66.4 (OCH₂), 75.7 (OCH), 109.9 (C), 135.7 (CH), 163.5 (CH), 208.7 (CO); HRMS (ESI) calcd for C₁₃H₁₈O₃Na (M + Na)⁺, 245.1154; found, 245.1179.

Following the procedure for the reduction of the cyclopentenone derivative **8a**, the cyclopentenone derivative **8b** (310 mg, 1.40 mmol) was reduced with LiAlH₄ (58 mg, 1.54 mmol) at -60 °C to the cyclopentenol derivative **9b** (272 mg, 75%) and its C-1 diastereoisomer (18 mg, 5%) as viscous liquid.

9b. $R_f = 0.55$ (EtOAc/petroleum ether 1:1); $[\alpha]^{24}{}_D + 33.0$ (*c* 1.7, CHCl₃); ¹H NMR δ 1.32–1.38 (2H, m, CH₂), 1.48–1.66 (8H, m, CH₂), 2.21 (2H, ddd, J = 7.2, 8.2, 14.1 Hz, CH₂), 2.40–2.42 (1H, m, OH), 2.78–2.82 (1H, m, CH), 3.57 (1H, t, J = 7.9 Hz, OCH₂), 4.03 (1H, dd, J = 6.5, 7.9 Hz, OCH₂), 4.12–4.18 (1H, m, OCH), 4.61–4.63 (1H, m, C₁–H), 5.80 (1H, dd, J = 2.2, 5.6 Hz, =CH), 5.95–5.98 (1H, m, =CH); ¹³C NMR δ 23.6 (CH₂), 23.9 (CH₂), 25.1 (CH₂), 33.9 (CH₂), 34.6 (CH₂), 35.8 (CH₂), 46.2 (CH), 67.7 (OCH₂), 75.3 (OCH), 75.5 (OCH), 109.8 (C), 134.4 (CH), 135.5 (CH); HRMS (ESI) calcd for C₁₃H₂₀O₃Na (M + Na)⁺, 247.1310; found, 247.1299.

C-1 Diastereomer of 9b. $R_f = 0.50$ (EtOAc/petroleum ether 1:1); [α]²⁴_D +131.8 (*c* 2.3, CHCl₃); ¹H NMR δ 1.33–1.40 (2H, m, CH₂), 1.48–1.60 (8H, m, CH₂), 1.84 (1H, ddd, J = 3.2, 8.0, 11.2 Hz, CH₂), 2.08 (1H, ddd, J = 4.7, 7.2, 12.0 Hz, CH₂), 3.06–3.12 (2H, m), 3.51–3.58 (1H, m), 3.94–4.01 (2H, m), 4.87–4.90 (1H, m, C₁–H), 5.81 (1H, dd, J = 1.9, 5.5 Hz, =CH), 5.92–5.95 (1H, m, =CH); ¹³C NMR δ 23.7 (CH₂), 23.9 (CH₂), 25.1 (CH₂), 34.8 (CH₂), 36.2 (CH₂), 36.4 (CH₂), 47.9 (CH), 67.0 (OCH₂), 76.9 (OCH), 77.9 (OCH), 109.5 (C), 134.9 (=CH) 135.7 (=CH); HRMS (ESI) calcd for C₁₃H₂₀O₃Na (M + Na)⁺, 247.1310; found, 247.1333.

(1S,2R,3R,4S)-2-(Benzyloxy)-4-[(2S)-1,4-dioxaspiro[4.5]dec-2yl]cyclopentane-1,3-diol 29a and (1R,2R,3R,4S)-2-(Benzyloxy)-4-[(2S)-1,4-dioxaspiro[4.5]dec-2-yl]cyclopentane-1,3-diol 30a. To a magnetically stirred solution of BF₃·OEt₂ (5 mL) in diglyme (20 mL) at 45 °C was added NaBH₄ (1.0 g) in small portion, and the BH₃ gas thus liberated was subsequently bubbled through a solution of the cyclopentenone derivative 27 (330 mg, 1.01 mmol) in THF (20 mL) at -10 °C till complete disappearance of the starting cyclopentenone derivative. The reaction mixture was then warmed to 20 °C and was diluted with THF (10 mL). To the magnetically stirred reaction mixture, cooled to -10 °C was added dropwise aqueous NaOH solution (3 N, 7 mL) to make it alkaline (pH paper) followed by H₂O₂ (30%, 7 mL). The reaction mixture was allowed to attain room temperature very slowly when the two layers separated. The organic layer was separated, and the aqueous layer was extracted with ether $(2 \times 10 \text{ mL})$. The two organic layers were combined and washed with water (4 \times 2 mL) and brine (2 \times 2 mL) and dried, and the solvent was removed in vacuo. The residual mass was column chromatographed using ethyl acetate/petroleum ether (3:7) as eluent to afford the alcohols 29a (70 mg, 20%) and **30a** (190 mg, 54%).

29a. $[\alpha]^{23}{}_{\rm D} - 5.1$ (*c* 0.3, CHCl₃); ¹H NMR δ 1.39–1.40 (2H, m, CH₂), 1.48–1.71 (8H, m, CH₂), 2.06–2.22 (3H, m, CH₂, CH), 2.98 (2H, br s, OH), 3.61 (1H, dd, *J* = 4.9, 6.5 Hz), 3.75 (1H, dd, *J* = 4.4, 6.3 Hz), 3.97–4.07 (3H, m), 4.11 (1H, dd, *J* = 4.4, 10.5 Hz), 4.65 (1H, d, *J* = 11.8 Hz, PhCH₂), 4.82 (1H, d, *J* = 11.8 Hz, PhCH₂), 7.26–7.39 (5H, m, ArH); ¹³C NMR δ 23.8 (CH₂), 24.1 (CH₂), 25.0 (CH₂), 32.2 (CH₂), 35.0 (CH₂), 36.3 (CH₂), 45.8 (CH), 68.0 (OCH₂), 72.1 (OCH₂), 74.4 (OCH), 79.4 (OCH), 80.1 (OCH), 91.3 (OCH), 109.9 (C), 127.7 (CH), 127.8 (CH), 128.4 (CH), 138.4 (C); HRMS (ESI) calcd for C₂₀H₂₈O₅Na (M + Na)⁺, 371.1834; found, 371.1846.

30a. $[\alpha]^{23}_{\text{D}} - 1.9$ (*c* 0.8, CHCl₃); ¹H NMR δ 1.30–1.40 (2H, m, CH₂), 1.53–1.64 (8H, m, CH₂), 1.80–1.86 (2H, m, CH₂), 2.01–2.11 (1H, m, CH), 2.68 (1H, br s, OH), 2.78 (1H, br s, OH), 3.61 (1H, dd, *J* = 6.2, 7.8 Hz), 3.70 (1H, dd, *J* = 5.5, 7.2 Hz), 4.02–4.15 (3H, m), 4.25 (1H, t, *J* = 7.7 Hz), 4.68 (1H, d, *J* = 11.8 Hz, PhCH₂), 4.81 (1H, d, *J* = 11.7 Hz, PhCH₂), 7.26–7.38 (5H, m, ArH); ¹³C NMR δ 23.8 (CH₂), 24.1 (CH₂), 25.0 (CH₂), 31.7 (CH₂), 34.9 (CH₂), 36.4 (CH₂), 44.8 (CH), 67.9 (OCH₂), 68.9 (OCH), 72.2 (OCH₂), 79.3 (OCH), 79.9 (OCH), 85.0 (OCH), 109.9 (C), 127.9 (CH), 128.5 (CH), 137.9 (C); HRMS (ESI) calcd for C₂₀H₂₉O₅ (M + H)⁺, 349.2015; found, 349.2029.

(4R,6S)-6-[(2S)-1,4-Dioxa-spiro[4.5]dec-2-yl]octa-1,7-dien-4ol and Its C-4 Diastereoisomer 33. A solution of allyl bromide (1.0 g, 8.37 mmol) in THF (10 mL) was added dropwise to a stirred suspension of commercial zinc dust (0.4 g, 6.69 mmol) in THF (5 mL) at room temperature, and the mixture was stirred for 1 h to produce a clear solution. A solution of the aldehyde **3b** (1.25 g, 5.58 mmol) in THF (15 mL) was added dropwise to this solution at room temperature and stirred for 1 h. The reaction mixture was then cooled to 0 °C and quenched with a few drops of water. Usual workup of the reaction mixture followed by column chromatography using ether/petroleum ether (1:6) as eluent afforded an inseparable mixtures of dienols **33** (1.23 g, 83%) in a ca. 1:1 ratio: ¹H NMR δ (for both diastereoisomers) 1.35 (2H, br s), 1.47–1.53 (1H, m), 1.58 (4H, br s), 1.62 (4H, br s), 1.72-1.82 (1H, m), 2.23-2.29 (2H, m), 2.33-2.40 (1H, m), 2.60 (1H, br s), 3.62-3.73 (2H, m), 3.85-3.97 (2H, m), 5.05-5.10 (4H, m), 5.52-5.59 (1H, m), 5.80-5.87 (1H, m); ¹³C NMR δ (for both isomers) 23.9 (CH₂), 24.0 (CH₂), 24.1 (CH₂), 25.10 (CH₂), 25.15 (CH₂), 35.2 (CH₂), 35.3 (CH₂), 36.4 (CH₂), 36.5 (CH₂), 39.7 (CH₂), 39.8 (CH₂), 41.4 (CH₂), 42.68 (CH₂), 42.70 (CH₂), 45.6 (CH), 46.3 (CH), 68.1 (CH₂), 68.2 (CH₂), 68.9 (CH), 69.1 (CH), 78.0 (CH), 78.3 (CH), 109.9 (C), 110.0 (C), 117.0 (CH₂), 117.2 (CH₂), 117.5 (CH₂), 117.6 (CH₂), 135.0 (CH), 135.1 (CH), 138.0 (CH), 138.2 (CH); HRMS (ESI) calcd for $C_{16}H_{26}O_3Na$ (M + Na)⁺, 289.1780; found, 289.1779.

(5*R*)-5-[(2*S*)-1,4-Dioxa-spiro[4.5]dec-2-yl]-cyclohex-3-enone 34c. A solution of diastereoisomeric dienols 33 (1.1 g, 4.14 mmol) in dichloromethane (50 mL) was treated with Grubbs' first generation catalyst (90 mg, 0.11 mmol) under argon atmosphere and stirred at room temperature for 3 h. The residual mass obtained after removal of solvent was chromatographed using ether/petroleum

ether (1:4) to afford an inseparable mixture of cyclohexenols 34a and 34b (0.88 g, 89%) in a ca. 1:1 ratio.

A solution of cyclohexenols **34a** and **34b** (0.4 g, 1.68 mmol) was oxidized through Swern oxidation following the procedure for the synthesis of the aldehyde **3b** with oxalyl chloride (0.23 mL, 2.69 mmol), DMSO (0.36 mL, 5.04 mmol), and triethylamine (1.4 mL, 10.08 mmol) to afford the cyclohexenone **34c** (280 mg, 71%) as a colorless oil: IR (neat) ν_{max} 1718 cm⁻¹; ¹H NMR δ 1.36 (2H, br s), 1.51 (4H, br s), 1.54–1.59 (4H, m), 2.44–2.61 (2H, m), 2.74–2.78 (1H, m), 2.85–2.90 (2H, m), 3.62–3.69 (1H, m), 3.98–4.05 (2H, m), 5.74–5.79 (1H, m), 5.84–5.91 (1H, m); ¹³C NMR δ 23.7 (CH₂), 23.9 (CH₂), 25.1 (CH₂), 34.6 (CH₂), 35.9 (CH₂), 39.8 (CH₂), 40.4 (CH), 40.9 (CH₂), 66.6 (CH₂), 77.5 (CH), 110.0 (C), 126.4 (CH), 127.4 (CH), 209.0 (CO); HRMS (ESI) calcd for C₁₄H₂₁O₃ (M + H)⁺, 237.1491; found, 237.1470.

Acetic Acid (1*R*,5*R*)-5-[(2*S*)-1,4-dioxa-spiro[4.5]dec-2-yl]-cyclohex-3-enyl Ester 34d. The cyclohexenone 34c (280 mg, 1.19 mmol) was reduced with LiAIH₄ (50 mg, 1.30 mmol) at -60 °C, following the procedure for the reduction of 8a, to afford exclusively the cyclohexenol 34a (260 mg, 92%) as a colorless viscous liquid: $[\alpha]^{25}_{D} + 47.9$ (*c* 0.9, CHCl₃); ¹H NMR δ 1.22–1.37 (3H, m), 1.51 (4H, br s), 1.55–1.60 (4H, m), 1.97–2.00 (1H, m), 2.05–2.09 (1H, m), 2.33–2.38 (2H, m), 2.51 (1H, br s), 3.60–3.66 (1H, m), 3.88–3.91 (1H, m), 3.94–4.05 (2H, m), 5.45–5.51 (1H, m), 5.66–5.73 (1H, m); ¹³C NMR δ 23.7 (CH₂), 23.9 (CH₂), 25.1 (CH₂), 34.7 (CH₂), 34.8 (CH₂), 34.9 (CH₂), 36.0 (CH₂), 39.3 (CH), 66.4 (CH₂), 66.8 (CH), 78.1 (CH), 109.4 (C), 126.3 (CH), 126.8 (CH); HRMS (ESI) calcd for C₁₄H₂₂O₃Na (M + Na)⁺, 261.1467; found, 261.1487.

To a magnetically stirred solution of the cyclohexenol 34a (260 mg, 1.09 mmol) in dichloromethane (10 mL) at 0 °C was added dropwise acetic anhydride (0.17 mL, 1.86 mmol), triethyl amine (0.34 mL, 2.40 mmol), and 4-(dimethylamino)pyridine (4-DMAP; 10 mg). The reaction mixture was then allowed to stir under argon atmosphere for 2 h at room temperature. Solvent was evaporated, and the residual mass was extracted with ether (30 mL). The ether layer was then washed with water (2 mL) followed by brine (2 mL) and dried. Removal of solvent under reduced pressure followed by column chromatography using ether/petroleum ether (1:9) afforded the corresponding acetate 34d (280 mg, 92%) as a colorless oil: $[\alpha]^{25}_{D}$ +53.6 (c 1.8, CHCl₃); IR(neat): ν_{max} 1732 cm⁻¹; ¹H NMR δ 1.24–1.43 (3H, m), 1.56 (4H, br s), 1.68 (4H, br s), 2.02 (3H, s, -OCOCH₃), 2.03-2.15 (2H, m), 2.39-2.44 (1H, m), 2.52-2.58 (1H, m), 3.61-3.68 (1H, m), 3.92-4.03 (2H, m), 4.92-5.01 (1H, m), 5.46–5.49 (1H, m), 5.68–5.73 (1H, m); $^{13}\mathrm{C}$ NMR δ 21.3 (CH₃), 23.8 (CH₂), 23.9 (CH₂), 25.1 (CH₂), 31.2 (CH₂), 31.3 (CH₂), 34.8 (CH₂), 36.1 (CH₂), 39.3 (CH), 66.6 (CH₂), 69.8 (CH), 78.1 (CH), 109.4 (C), 126.3 (CH), 126.5 (CH), 170.5 (CO); HRMS (ESI) calcd for $C_{16}H_{24}O_4Na (M + Na)^+$, 303.1572; found, 303.1528.

Epoxidation of Acetate 34d. To a magnetically stirred cold (0 °C) solution of the acetate **34d** (170 mg, 0.61 mmol) in 1,2dichloroethane (4.5 mL) was added *m*-CPBA (209 mg, 1.21 mmol) in one portion. After stirring at room temperature for 8 h, the reaction mixture was cooled to 0 °C and quenched by adding saturated Na₂SO₃ solution, extracted with ether (30 mL), and washed with ice-cold 2% NaOH solution (w/v) (3 × 2 mL) and dried. Evaporation of the solvent followed by column chromatog-raphy using ether/petroleum ether (1:6) afforded an inseparable mixture (2:3) of epoxides **35** and **36** (160 mg, 89%) as a colorless viscous liquid: IR(neat) ν_{max} 1728 cm⁻¹; ¹H NMR δ (for mixture of diastereomers) 1.23–1.42 (6H, m), 1.59 (8H, br s), 1.62 (8H, br s), 1.70–1.84 (4H, m), 2.02 (3H, s, CH₃ of one isomer), 2.03 (3H, s, CH₃ of other isomer), 2.23–2.54 (4H, m), 2.99–3.01 (1H, m), 3.07–3.08 (1H, m), 3.12–3.19 (1H, m), 3.29–3.31 (1H, m), 3.73 (1H, dd, J = 7.5, 7.2 Hz), 3.87–3.92 (1H, m), 4.06–4.14 (3H, m), 4.23 (1H, dd, J = 12.6, 6.6 Hz), 4.67–4.73 (1H, m), 4.83–4.91 (1H, m); ¹³C NMR (for both isomers) δ 21.1 (CH₃), 21.2 (CH₃), 23.7 (CH₂), 23.8 (CH₂), 23.9 (CH₂), 25.1 (CH₂), 27.0 (CH₂), 28.8 (CH₂), 29.4 (CH₂), 30.6 (CH₂), 34.5 (CH₂), 35.1 (CH₂), 36.0 (CH₂), 36.2 (CH₂), 37.6 (CH), 38.2 (CH), 49.7 (CH), 51.8 (CH), 52.8 (CH), 53.8 (CH), 66.6 (CH₂), 66.7 (CH₂), 67.8 (CH), 68.6 (CH), 76.6 (CH), 77.2 (CH), 109.4 (C), 109.8 (C), 170.1 (CO), 170.4 (CO); HRMS (ESI) calcd for C₁₆H₂₄O₅Na (M + Na)⁺, 319.1521; found, 319.1528.

(1R,3S,5S,6S)-5-[(2S)-1,4-Dioxa-spiro[4.5]dec-2-yl]-7-oxabicyclo[4.1.0]heptan-3-ol 37 and (1R,4S,5R)-5-[(2S)-1,4-dioxaspiro[4.5]dec-2-yl]-cyclohex-2-ene-1,4-diol 38a. To a magnetically stirred solution of diphenyl diselenide (125 mg, 0.4 mmol) in absolute ethanol (1.5 mL) was added sodium borohydride (26 mg, 0.69 mmol) in portions, and stirring was continued until the bright yellow solution turned colorless. A solution of epoxides 35 and 36 (170 mg, 0.57 mmol) in absolute ethanol (1.5 mL) was added dropwise to the reaction mixture, and the reaction mixture was stirred at room temperature for 2 h. The reaction mixture was cooled to 0 °C, and THF (2.5 mL) was added followed by addition of H₂O₂ (30%) (0.6 mL) over a period of 3 min. The reaction mixture was left to attain room temperature and then refluxed at 70 °C for 1.5 h. The resulting slurry was diluted with water (2 mL) and extracted with ether $(3 \times 5 \text{ mL})$. The organic phase was washed with aqueous Na₂CO₃ (10%, 2×2 mL), dried, and concentrated. Column chromatography of the residual mass using ethyl acetate/ petroleum ether (3:5) afforded the epoxy alcohol 37 (42 mg, 29%) as a colorless viscous liquid: IR(neat) ν_{max} 3421 cm⁻¹; ¹H NMR δ 1.40 (2H, br s), 1.50–1.63 (9H, m), 1.77–1.81 (1H, m), 2.24– 2.36 (2H, m), 2.37-2.48 (2H, m), 3.00-3.01 (1H, m), 3.29-3.30 (1H, m), 3.71-3.76 (1H, m), 3.95-4.02 (2H, m), 4.07-4.12 (1H, m); ¹³C NMR δ 23.7 (CH₂), 23.9 (CH₂), 25.1 (CH₂), 34.2 (CH₂), 34.5 (CH₂), 36.0 (CH₂), 36.7 (CH₂), 37.5 (CH), 53.1 (CH), 54.4 (CH), 64.5 (CH), 66.6 (CH₂), 76.6 (CH), 109.9 (C); HRMS (ESI) calcd for $C_{14}H_{22}O_4Na (M + Na)^+$, 277.1416; found, 277.1439. The cyclohexenediol 38a (60 mg, 41%) was also produced as a white crystalline solid: mp 142 °C; $[\alpha]^{25}_{D}$ +32.1 (*c* 0.5, CHCl₃); IR (KBr) $\nu_{\rm max}$ 3246.0 cm⁻¹; ¹H NMR δ 1.40 (2H, br s), 1.59 (4H, br s), 1.61 (4H, br s), 1.65-1.68 (2H, m), 2.11-2.17 (1H, m), 2.45 (1H, br s), 2.54 (1H, br s), 3.78 (1H, dd, J = 7.0, 6.8 Hz), 4.05–4.07 (1H, m), 4.10–4.30 (3H, m), 5.82–5.91 (2H, m); $^{13}\mathrm{C}$ NMR δ 23.8 (CH₂), 24.0 (CH₂), 25.1 (CH₂), 29.0 (CH₂), 34.8 (CH₂), 36.3 (CH₂), 41.1 (CH), 64.9 (CH), 67.0 (CH₂), 67.5 (CH), 76.4 (CH), 109.5 (C), 129.1 (CH), 135.1 (CH); HRMS (ESI) calcd for C₁₄H₂₂O₄Na $(M + Na)^+$, 277.1416; found, 277.1461.

Acknowledgment. Financial support from the Department of Science and Technology, Government of India is gratefully acknowledged. T.B. and N.S. wish to thank CSIR, New Delhi for Senior Research Fellowships.

Supporting Information Available: General methods and procedures along with spectral data for compounds **3b**, **10b**, **17**–**25**, **27**, **29b**, **30b**, **31**, **32**, **38b**, **39a**,**b**, and **40a**–**c** and copies of ¹H, ¹³C NMR and DEPT spectra for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

JO061717T